Comparative Study of the Nutritional Compositions of Pure and Adulterated Honey Samples Collected from Kano South Senatorial District of Kano State, Nigeria

Datti Ya'u, Usman Lado Ali and Ummi Umar Ahmad

ABSTRACT

Honey, the sweet, nutritious substance produced by honeybees, is of good nutritional and medicinal importance to consumers. The present study aimed at comparing the nutritional compositions of pure and adulterated honey samples collected directly from the honey beekeepers from Kano South Senatorial District of Kano State, Nigeria. For this purpose, a total of three samples each were randomly collected from three randomly selected local governments of the Senatorial District and analyzed for some nutritional values. Similarly, an adulterated honey sample was randomly collected from each of the three local governments. The mean nutritional values obtained for the pure sample were as follows: Moisture content (17.69-18.02%); crude fat (0.89-1.29%); crude protein (0.02-0.13%); ash content (0.44-0.51%); carbohydrate content (80.16-80.75%); energy values (1386.613-1396.157 KJ/100 g). While the adulterated honey sample recorded the following results: Moisture content (24.01%); crude fat (0.94%); crude protein (1.26%); ash content (2.64%); carbohydrate content (71.15%); energy values (1247.847 KJ/100 g). These results indicate that the pure honey sample analyzed are of good quality, as such could be used for both nutritional and medicinal purposes, having met the international honey standards. On the other hand, the adulterated honeys should always be used with care since they usually fall below the international honey standards.

Published Online: December 22, 2020

ISSN: 2684-4478

DOI:10.24018/ejchem.2020.1.6.35

Datti Va'u*

Department of Chemistry, Yusuf Maitama Sule University, Kano State, Nigeria.

(e-mail: whydee@yahoo.co.uk)

Usman Lado Ali

Department of Chemistry, Yusuf Maitama Sule University, Kano State, Nigeria.

Ummi Umar Ahmad

Department of Science Laboratory Technology, School of Technology, Kano, Nigeria.

*Corresponding Author

Keywords: Adulteration; Honey; Kano South; Nutritional Composition.

I. INTRODUCTION

Honeybees (*Apis mellifera*) are the common producers of honey from the nectar of plants or from honeydew [1]-[3]. Honey is a unique, natural, liquid sweetening agent whose composition mainly varies depending on its environmental and geographical location, as well as climatic conditions, the plant species, the methods employed during collection and storage [4]-[6].

Sugars are the principal components of honeys, with other nutritional components including some varieties of proteins, enzyme, vitamins, hormones, minerals, some heavy metals, yeast, as well as organic acids (mainly pyruvic, lactic, and formic acids) are also present [3], [7]-[11]. With the analysis of these parameters of honeys serving as quality indicator of individual honeys [6], [12].

Principally, honey is widely used as food and as sweetening agent, but it is reported to have many medicinal applications [13], [14], exhibiting many healing powers [15], [16], as well as antioxidant, anti-inflammatory, antifungal, antimicrobial, and antibacterial activities [16]-[20].

Many researches to determine the nutritional composition and physicochemical properties of honeys samples have been conducted and reported [3], [6], [12], [21]-[31]. With more researches are still on-going, because knowledge of the physicochemical properties, the nutritional compositions and the possible toxicological effects of all the locally available honeys is paramount importance, especially it helps in eliminating all dietary problems associate with either deficiency or consumption of such honey [32], [33].

It is in line of this that this study was carried out to analyze and compare the nutritional composition of pure and adulterated honey samples collected from Kano South Senatorial District.

II. MATERIALS AND METHODS

A. Samples Collection

Three samples of both pure and adulterated honey samples were directly collected from different honey beekeepers from each of the three randomly selected local governments from the Kano South Senatorial District. Similarly, adulterated honey samples were randomly collected from the markets of each of the three randomly selected local governments. In order to avoid the growth of microorganisms as well as absorption of moisture, all the collected samples were separately stored in clean, labelled, airtight bottles at an ambient temperature [6], [16]. The samples were collected from the following local government areas of the Kano Senatorial District; Bebeji, Kibiya and Ajingi local

government areas.

B. Determination of Moisture Content

The moisture contents of the honey samples were analyzed in accordance with the procedure by Association of Official Analytical Chemists [34]. Two grams each of the honey samples were weighed and then transferred into pre-weighed crucibles, then placed into an oven at 105 °C for 5 hours. Each crucible and its content were then removed from the oven and then cooled in a desiccator, then re-weighed. Triplicate results were obtained for each sample and the mean value was reported to three decimal points according to the following formula:

Moisture Contents (%) =
$$\frac{W1 - W2}{W1} \times 100\%$$

where

W1= Sample weight before drying.

W2 = Sample weight after drying.

C. Determination of Crude Fat Content

The fat contents of the honey samples were analyzed according to the AOAC [34], as adopted by Ibe et al. [35]. In this procedure, 2 g of each of the honey samples was weighed into a clean dry beaker, and then concentrated HCl was added, followed by petroleum ether, in order to digest the sample. The chemicals and the honey samples were separated using a separation funnel, and the floatable part of the mixture was collected and transferred into a separate beaker and then heated for 5 minutes to dryness. After drying, the residue was then oven-dried then weighed, with the fat content calculated using the formula below:

Fat Content (%) =
$$\frac{(M_2 - M_1)}{M_3} \times 100$$

where

 $M_1 = Mass of the flask.$

 $M_2 = Mass of flask + fat.$

 M_3 = Mass of the sample.

D. Determination of Crude Protein Content

The protein contents of all the honey samples were analyzed using Kjeldahl method according to the AOAC [34] procedure, as adopted by Adeniyi et al. [30]. In this method, the nitrogen content of each sample was first determined, and then used to estimate the protein content using the 6.25 conversion factor.

E. Determination of Ash Content

The ash contents of each of the honey samples were determined according to the method according to the AOAC [34] procedure, as adopted by Adeniyi et al. [30]. Here 5 g of each of the honey sample was transferred into a pre-weighed, tarred porcelain crucible and the placed into a muffle furnace (Model Box 1200) at 550 °C until a white-gray ash was obtained. The crucible was then brought out and then transferred to a desiccator where it was cool to room temperature and then reweighed. The ash contents of all the honey samples were then calculated as a percentage based on the initial weight of the sample using the following formula:

$$\begin{array}{l} \textit{Ash Contents (\%) =} \\ = \frac{(\textit{Weight of Crucible} + \textit{Ash}) - (\textit{Weight of Crucible})}{\textit{Initial Weight of the Sample}} \times 100\% \\ \end{array}$$

F. Determination of the Carbohydrate

The carbohydrate contents of all the honey samples were obtained using the following formula as reported by Adeniyi et al. [30] and Oyeyemi et al. [36].

G. Determination of the Energy Values

The energy values of the samples were determined using the following formula as reported by Adeniyi et al. [30] and Buba et al. [37].

Energy (KJ/100 g) =
$$4.186[(\% Crude Protein \times 4) + (\% Crude Fat \times 9) + (\% Carbohydrate \times 4)]$$

III. RESULTS AND DISCUSSION

The nutritional compositions of all the honey samples analyzed are presented in the Table 1 below.

TABLE 1: NUTRITIONAL O	COMPOSITIONS OF	F THE HONEY SAMPLES
------------------------	-----------------	---------------------

TABLE 1. NUTRITIONAL COMPOSITIONS OF THE HONET SAMPLES							
Sample	Moisture	Crude Fat	Crude	Ash Content	Carbohydrate	Energy Values	
	Content (%)	(%)	Protein (%)	(%)	Content (%)	KJ/100 g	
KS A	17.83	0.89	0.06	0.47	80.75	1386.613	
KS B	17.69	1.21	0.13	0.44	80.53	1396.157	
KS C	18.02	1.29	0.02	0.51	80.16	1391.133	
KS AH	24.01	0.94	1.26	2.64	71.15	1247.847	
WHO/	< 20	0-10	0.1-1.5	< 0.6	75-85		
FAO	≥ 20	0-10	0.1-1.3	≥ 0.0	13-83		

Key: KS A = Bebeji; KS B = Ajingi; KS C = Kibiya; KS AH = Adulterated Honey.

Moisture content is one of the most important parameters in evaluating the quality of honey [38]. A moisture content higher than the range of <20%, the international standard, might increase the chance of fermentation [39] and possible attacks by microorganisms [40], as well as reduces the honey's shelf life [41]. In this study, the moisture contents of all the three pure honey samples were found to be within the internationally accepted range of <20%. The moisture contents of the samples ranged between 17.69% (Ajingi sample) to 18.02% (Kibiya sample), indicating that the results are in agreement with the Codex Alimentarius Standard for honey [42]. The results are also in agreement with similar reports by Laleh et al. [6]; Omafuvbe and Akanbi [17]; Boussaid et al. [31]; Ahed and Khalil [38]; Navar et al. [40]; Atrouse et al. [43]; Malika et al. [44]; Ibrahim Khalil et al. [45] and Eleazu et al. [46]. However, the moisture of the adulterated honey sample was found to be 24.01% and this is an indication of adulteration [47]. And according to El-Bialee and Sorour [2], moisture content increase ranging from 20.7 to 39.6% could be due to adulteration by starch, glucose, or

Fats, one of the three main macronutrients, are a necessary part of the diet of both humans and animals, and the most efficient form of energy storage [48]. They are mainly present in honey as sterols, glycerides and sometimes as phospholipids. Low fat level in honey is an indication of its virginity [12]. The fat contents of the honey samples investigated in this research fall within the range of 0.89% (Bebeji sample) to 1.29% (Kibiya sample). Low fat content reported in this study is supported by similar reports from other researchers [36] [37] [49] [50] [51] [52] whose reports showed little or no fat in honey samples. This indicates that honey cannot be considered a good source of fat. The result from the adulterated honey sample was also found to be low, and this may possibly be attributed to the fact that the adulterants used did not contain fatty substances. However, both the pure and adulterated honey samples have recorded fat contents within the 0 to 10% range of fat content recommended by the World Health Organization.

Just like the fat content, very low value was also recorded for the protein content. The World Health Organization has set a standard for protein content to be between the range of 0.1 to 1.5%, and this is an indication that honey is not an adequate source of protein [30]. The protein contents obtained in this study are between the range of 0.02% (Kibiya sample) to 0.13% (Ajingi sample), and these are within the WHO standards, and also in agreement with similar report by Buba et al. [37] who reported the protein content of honey in North-East of Nigeria as 0.35 to 1.08%. Other researchers [30], [31], [38], [53]-[57] have also reported similar results. Although most honey proteins are enzymes added by bees, a much higher protein content could be an indication of high pollen content which indicates natural, good-quality honey [38] [56]. But unlike for the fats contents, the results for the protein content from the adulterated honey sample were also found to be a little bit high (though below the WHO maximum limit), and this may possibly be attributed to the fact that the adulterants used might have been proteincontaining substances.

Ash contents depend on the mineral content of the honey and it is the measure of the inorganic residue of the honey [58]. The ash contents of all the honey samples analyzed were found to be between the range of 0.44% (Ajingi sample) and 0.51% (Kibiya sample), and these results are supported by similar previous results by Adenekan et al. [59] who reported ash contents in the range of 0.12 to 0.50% for honey samples; Gulfaraz et al. [41] who reported a range of 0.13 to 0.44% ash contents for various honey types of Pakistan; Buba et al. [37] who reported a range of 0.37 to 0.54% for some Nigerian honey samples; Oyeyemi et al. [36] who reported a range of 0.44 to 0.58% for honey samples obtained from Ekiti State, Nigeria. The results obtained in this study are however a little below 0.70 to 1.67% ash contents reported by Kambai et al. [60] for honey samples from selected hives in Jos Metropolis, Nigeria. However, the Codex Alimentarius Commission [61] standard, as well as the WHO specified an ash content of not more than 0.6% for normal honey. The 2.64% ash content recorded for the adulterated honey sample is a clear indication of adulteration, as adulteration of honey is reported to increase its ash content [62], [63]. The Ash content is an important parameter used in determining the floral origin of honeys. Even though according to the Codex Alimentarius Standards, the ash content of pure honey must fall below 0.6%, it is reported that blossom honeys have mineral contents mostly between 0.1 and 0.3%, while that of honeydew honeys can reach up to 0.6% [9].

Honey carbohydrate mainly includes a complex mixture of monosaccharides (glucose and fructose) up to about 70%, then about 10% of disaccharides, and then small amounts of other higher sugars [58]. The results of the carbohydrate contents of the honey samples analyzed in this study were found to be within the range of 80.16% (Kibiya sample) to 80.75% (Bebeji sample), and these are within the WHO standard of 75 to 85%. The results from this study are in agreement with similar reports by other researchers [30], [35], [37], [64]-[66]. Glucose and fructose are the major components of carbohydrate found in honey and their ratio is a factor in determining the adulteration levels and the honeys suitability in managing cases of diabetes [67], [68]. The low carbohydrate content (71.15%) of the adulterated honey sample might possibly be due to adulteration with non-sugar substances like water [69].

Honey is regarded as an important energy-providing food throughout the world, with the energy value of honey being an important quality that depends on the presence of simple, inverted sugars, such as glucose and fructose, which are an immediate and prompt source of energy for the human body [70], [71]. The mean energy values of all the honey samples analyzed in this study were found to be between 1386.613 KJ/100 g (Bebeji sample) to 1396.157 KJ/100 g (Ajingi sample). These results are in agreement with similar results by other researchers: 329.12 to 333.64 Kcal/100 g reported by Adeniyi et al. [30]; 281.45 to 507.16 Kcal/100 g reported by Oyeyemi et al. [36]; 1383.23 to 1410.20 KJ/100 g reported by Buba et al. [37]; 326.2 to 328.06 Kcal/100 g reported by Ndife et al. [66]; 303 Kcal/100 g reported by Blassa et al. [72]; 1383.23 to 1410.20 KJ/100 g reported by Amabye and Frehiwot [73]. The energy values obtained in the present study is above the WHO minimum limit of 1228 KJ/100 g. Honey provides enough calories (64 calories per tablespoon), from which it derives its energy value [13]. The energy values in honey samples could therefore be an important source of dietary calories. Honey contains high energy carbohydrate foods which are easily metabolized by the body as against other available refined sugar, and this is why honey is a recommended food for both children and adults of all sexes [36]. The adulterated honey sample recorded a much lower energy value of 1247.847 KJ/100 g, and this might have been due to the fact the adulteration was done with non-energy producing substances [69].

IV. CONCLUSION

This study indicated that pure honey samples contain similar physico-chemical properties, with such properties easily affected during adulteration. In general, this study revealed that the analyzed pure honey samples from Kano North Senatorial District are essential for the human diet,

growth, and health, as the levels of the physico-chemical properties were within the permissible limit set by the FAO/WHO. However, adulterated honey samples may not have that guarantee because some parameters exceed the permissible limit set by the FAO/WHO.

V. ACKNOWLEDGMENT

Our acknowledgment goes to the management of Yusuf Maitama Sule University, Kano who gave full financial, moral, and technical support to this research. We also wish to acknowledge the assistance and support we received from the Chemistry Department of Yusuf Maitama Sule University, Kano. We would also want to thank all the beekeepers who supplied the honey samples for this study.

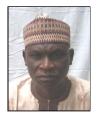
REFERENCES

- [1] Codex Alimentarius Commission Standards. (2002): Draft revised standard for honey. 9-11 February 2000. Joint FAO/WHO Food Standards Programme.
- El-Bialee NM. and Sorour MA. (2011): Effect of Adulteration on Honey Properties, International Journal of Applied Science and Technology, 1(6):122-133.
- Adugna E., Ariaya H., Gebremariam B. and Ayenew A. (2020): Determination of some Heavy Metals in Honey from Different Regions of Ethiopia, Cogent Food and Agriculture, 6:1-14.
- De Rodriguez GO., de Ferrer BS., Ferrer A. and Rodriguez B. (2004): Characterization of Honey Produced in Venezuela. Food Chemistry, 84(4):499-502.
- Kucuk M., Kolaylı S., Karaoglu S., Ulusoy E., Baltacı C. and Candan F. (2007): Biological Activities and Chemical Composition of Three Honeys of Different Types from Anatolia. Food Chemistry, 100(2):526-534.
- Laleh M., Mohsen E. and Ali H. (2013): Evaluation of some Physicochemical and Rheological Properties of Iranian Honeys and the Effect of Temperature on its Viscosity. American-Eurasian Journal of Agricultural and Environmental Sciences, 13(6):807-819.
- [7] Hernandez OM., Fraga JMG., Jimenez AI., Jimenez F. and Arias JJ. (2005): Characterization of Honey from the Canary Islands: Determination of the Mineral Content by Atomic Absorption Spectrophotometry. Food Chemistry, 93(3):449-458.
- Kujawski MW. and Namiesnik J. (2008): Challenges in Preparing Honey Samples for Chromatographic Determination of Contaminants and Trace Residues. TrAC Trends in Analytical Chemistry, 27(9):785-
- Bogdanov S. (2009): Honey Composition. In: Book of Honey, Chapter 5. Bee Product Science. Pp. 1-10.
- [10] Pohl P. (2009): Determination of Metal Content in Honey by Atomic Absorption and Emission Spectrometries. Trends in Analytical Chemistry, 28:117-128.
- [11] Wang J. and Li QX. (2011): Chemical Composition, Characterization, and Differentiation of Honey Botanical and Geographical Origins. In S. Taylor (Ed.), Advances in Food and Nutrition Research 62:89-137.
- [12] Juszczak L., Socha R., Roznowski J., Fortuna T. and Nalepka K. (2009): Physicochemical Properties and Quality Parameters of Herb Honeys. Food Chemistry, 113:538-542
- [13] Abeshu MA. and Geleta B. (2016): Medicinal Uses of Honey. Biology and Medicine (Aligarh). 8(2):01-07.
- [14] Farida I., Rashid M., Noor I., Ghulam S, Asif MM. and Hammad S. (2014): Physicochemical Analysis of Honey Samples Collected from Local Markets of Rawalpindi and Islamabad, Pakistan, American Journal of Biochemistry, 4(2):35-40.
- [15] Abdullah F. and Abdulaziz MA. (1998): The Prophylactic and Curative Effect of Cedar Honey Induced Ulcers in Rabbits. The Second International Arab Apiculture. Conference, Amman, 1:26-31.
- [16] Gulfraz M., Ifftikhar F., Asif S., Raja GK., Asad JM.1, Imran M., Abbasi K. and Zeenat A. (2010): Quality assessment and antimicrobial activity of various honey types of Pakistan, African Journal of Biotechnology 9(41):6902-6906.
- [17] Omafuvbe BO. and Akanbi OO. (2009): Microbiological and Physico-Chemical Properties of some Commercial Nigerian Honey. African Journal Microbiol. Res 3:891-896.

- [18] Gomes S., Dias LG., Moreira LL., Rodrigues P. and Estevinho L. (2010) Physico-Chemical, Microbiological and Antimicrobial Properties of Commercial Honeys from Portugal. Food Chemistry and Toxicology 48:544-548.
- [19] Kumar HMM., Anand AP., Deepa V., and Siddagangaiah H. (2013): Study of Physicochemical Parameters and Antioxidants in Honey Collected from Different Locations in India, International Journal of Pharmacy and Life Sciences, 4(12):3159-3165.
- [20] Datti Y., Ahmad UU. and Hafsat N. (2020): Comparative Analysis of the Mineral Compositions of Honey Samples Collected from the Three Senatorial Districts of Kano State, Nigeria. FUDMA Journal of Sciences (FJS), 4(3):170-177.
- [21] Martinez-Gomez ME., Guerra-Hernandez E., Montilla-Gomez JY. and Molins-Marin JL. (1993): Physicochemical Analysis of Spanish Commercial Eucalyptus honeys. Journal Apic. Res; 32(3/4):121-126.
- [22] Azeredo L., Da C., Azeredo MAA., de Souza SR. and Dutra VML. (2003): Protein Contents and Physicochemical Properties in Honey Samples of Apis mellifera of Different Floral Origins. Food Chemistry, 80:249-254.
- [23] Adebiyi FM., Akpan I., Obiajunwa EI. and Olaniyi HB. (2004): Chemical/Physical Characterization of Nigerian Honey. Pakistan Journal of Nutrition 3(5):278-281.
- [24] Rodriguez GO., Ferrer BS., Ferrer A. and Rodriguez B. (2004): Characterization of Honey Produced in Venezuela. Food Chemistry, 84:499-502
- [25] Downey G., Hussey K., Daniel-Kelly J., Walshe TF. and Martin PG. (2005): Preliminary Contribution to the Characterization of Artisanal Honey Produced on the Island of Ireland by Palynological and Physico-Chemical Data, Food Chemistry, 91(2):347-354.
- [26] Finola MS., Lasagno MC. and Marioli JM. (2007): Microbiological and Chemical Characterization of Honeys from Central Argentina. Food Chemistry, 100(4):1649-1653.
- [27] Al ML., Daniel D., Moise A., Bobis O., Laslo L., and Bogdanov S. (2009): Physico-Chemical and Bioactive Properties of Different Floral Origin Honeys from Romania. Food Chemistry, 112:863-867.
- [28] Ajlouni and Sujirapinyokul P. (2010): Hydroxymethylfurfuraldehyde and Amylase Contents in Australian Honey. Food Chemistry, 119(3):1000-1005.
- [29] Abel AA. and Adedoyin DB. (2011): Physico-Chemical Evaluation of the Authenticity of Honey Marketed in Southwestern Nigeria, Journal of Basic and Applied Scientific Research 1(12):3339-3344.
- [30] Adeniyi KA., Daudu OAY., Abubakar A., Ismail AD., Busari MB., Abdulsalami H. and Oyibo-Usman K A. (2014): Comprative Analysis of the Proximate and Nutritional Compositions of Nigerian Bitter and Sweet Honey from Apis mellifera, International Journal of Scientific and Research Publications, 4(11):1-4.
- [31] Boussaid A., Chouaibia M., Rezigb L., Hellalc R., Donsìa F., Ferraria G. and Hamdib S. (2018): Physicochemical and Bioactive Properties of Six Honey Samples from Various Floral Origins from Tunisia. Arabian Journal of Chemistry, 11(2):265-274.
- [32] Nnam NM. (2003). Evaluation of the nutrient and sensory properties of porridges from African yam bean (Sphenostylis stenocarpa) and maize (Zea mays L.) Flours, Nigeria Research in Education, 9:49-54.
- [33] Standing Committee on Nutrition (2006): The double burden of malnutrition - A challenge for cities worldwide, Third World Urban Forum-Vancouver 19-23 June 2006 SCN Statement.
- [34] AOAC. (2005) 18th Ed. Association of official Analytical Chemists.
- [35] Ibe AE., Onuoha GN., Adeyemi AA., Madukwe DK. and Udobi JO. (2013): Quantitative Analyses of Honey Samples from Four Different Sources in Abia State, Nigeria, International Journal of Natural and Applied Sciences, 9(2):107-116.
- [36] Oyeyemi SD., Kayode J. and Owolabi MO. (2015): Comparative Nutritional Studies on Honey Samples in Ado Ekiti, Ekiti State, Nigeria. Donnish Journal of Medicinal Plant Research 2(2):16-20.
- [37] Buba F., Gidado A. and Shugaba A. (2013): Analysis of Biochemical Composition of Honey Samples from North-East Nigeria. Biochemistry and Analytical Biochemistry 2(3):1-7.
- Ahed A. and Khalid MS. (2017): Physico-Chemical Properties of Multi-Floral Honey from the West Bank, Palestine, International Journal of Food Properties, 20(2):447-454.
- [39] Alvarez-Suarez JM., Tulipani S., Díaz D., Estevez Y., Romandini S., Giampieri F., Damiani E., Astolfi P., Bompadre S. and Battino M. (2010): Antioxidant and Antimicrobial Capacity of Several Monofloral Cuban Honeys and Their Correlation with Color, Polyphenol Content and other Chemical Compounds. Food and Chemical Toxicology, 48(8-9):2490-2499.
- [40] Nayar J., Shobhan K. and Kiran KC. (2017): Physicochemical Analysis of some Commercial Honey Samples from Telangana, Indian Journal of Nutrition, 4(1):01-04.

- [41] Gulfraz M., Iftikhar F., Imran M., Zeenat A., Asif S. and Shah I. (2011): Compositional Analysis and Antimicrobial Activity of Various Honey Types of Pakistan. International Journal of Food Science and Technology, 46(2):263-267.
- [42] Codex Alimentarius Standard for Honey (1993): Ref. Nr. CL 1993/14-SH FAO and WHO, Rome.
- [43] Atrouse OM., Oran SA, and Al-Abbadi SY, (2004): Chemical Analysis and Identification of Pollen Grains from Different Jordanian Honey Samples. International Journal of Food Science and Technology, 39(4):1-5.
- [44] Malika N, Mohamed F, Chakib E (2005). Microbiological and physico - chemical properties of Moroccan honey. Int. J. Agric. Biol., 7(5): 773-776.
- [45] Ibrahim Khalil MD., Mohammed M., Land B., Moukhtar B., Asiful Islam MD., Nazmul Islam MD., Siti AS. and Siew H. (2012): Physicochemical and Antioxidant Properties of Algerian Honey, Molecules 17:11199-11215.
- [46] Eleazu CO., Iroaganachi MA., Eleazu KC. and Okoronkwo JO. (2013): Determination of the Physico-Chemical Composition, Microbial Quality and Free Radical Scavenging Activities of Some Commercially Sold Honey Samples in Aba, Nigeria: "The Effect of Varying Colors." Journal of Nutrition Food Sciences, 4(1):32-41.
- [47] Nyau V., Mwanza P. and Moonga B. (2013): Physico-Chemical Qualities of Honey Harvested from Different Beehive Types in Zambia. African Journal of Food Agric. Nutrition and Development 13(2):7415-7427.
- [48] Pickova J. (2009): Importance of Knowledge on Lipid Composition of Foods to Support Development towards Consumption of Higher Levels of n-3 Fatty Acids via Freshwater Fish. Physiol. Res. 58(1):39-
- [49] Tan ST., Holand PT., Wilkins AL, and Molan PC. (1988): Extractives from New Zealand Honeys. Journal of Agricultural and Food Chemistry 36:453-460.
- [50] Singh N. and Kuar BP. (1997): Quality Evaluation of different Types of Indian Honey. Food Chemistry 58:129-133.
- [51] Khalil MI., Motallib MA., Anisuzzaman ASM., Sathi ZS. and Hye MA. (2001) Biochemical Analysis of Different Brands of Unifloral Honey Available at the Northern Region of Bangladesh. The Journal of Medical Sciences 1:385-388.
- [52] Blanka Z. and Lenka V. (2014): Adulteration of honey and available methods for detection - a review Acta Vet. Brno 83:85-102.
- [53] Anklam EA. (1998): Review of the Analytical Methods to Determine the Geographical and Botanical Origin of Honey. Food Chemistry 63(4):549-562.
- [54] Jagdish T. and Joseph I. (2004): Quantification of Saccharides in Multiple Floral Honeys Using Fourier Transform Infrared Micro-Attenuated Total Reflectance Spectroscopy. Journal of Agriculture and Food Chemistry 52(11):3237-3243.
- [55] Ouchemoukh S., Louailsche H. and Scheitzer P. (2007): Physicochemical Characteristics and Pollen Spectrum of Some Algerian Honeys. Food Control 18(1):52-58.
- [56] Nazarian H., Taghavizad R. and Majd A. (2010): Origin of Honey Proteins and Method for its Quality Control. Pakistan Journal of Botany 42(5):3221-3228.
- [57] Saxena S., Gautam S. and Sharma A. (2010): Physical, Biochemical, and Antioxidant Properties of Some Indian Honeys. Food Chemistry 118(2):391-397.
- [58] Teferi D. (2019): A Review on Effect of Adulteration on Honey Properties. Electronic Journal. Available SSRN: http://dx.doi.org/10.2139/ssrn.3359494.
- [59] Adenekan, M. O., Amusa, N. A. Lawal, A. O. and Okpeze, V. E. (2010): Physico-Chemical and Microbiological Properties of Honey Samples obtained from Ibadan, Journal of Microbiology and Antimicrobials, 2(8):100-104.
- [60] Kambai C., Popoola V., Ugbe CJ., Janfa N. and Ukanyirioha C. (2015): Proximate and Mineral Elements Composition of Honey from Selected Hives in Jos Metropolis, Nigeria, Journal of Environmental Science, Toxicology and Food Technology 9(9):94-98.
- [61] Codex Alimentarius Commission (2001): Revised Codex Standard for Honey, Codex STAN 12-1981, Rev.1 (1987), Rev.2; FAO/WHO, Pp. 215.
- [62] Abdel-Aal ESM., Ziena HM. and Youssef MM. (1993): Adulteration of Honey with High Fructose Corn Syrup: Detection by Different Methods. Food Chemistry, 48:209-212.
- [63] Ribeiro RD., Marsico OR., da Silva Carneiro ETC., Monteiro MLG, Junior C. C. and de Jesus EFO. (2014): Detection of Honey Adulteration of High Fructose Corn Syrup by Low Field Nuclear Magnetic Resonance (LF 1H NMR). Journal of Food Engineering, 135:39-43.

- [64] Doner L. (1997): The Sugars of Honey A Review. Journal Science Food Agriculture. 28:443-456.
- [65] Oyeleke SB., Dauda BEN., Jimoh T. and Musa SO. (2010): Nutritional Analysis and Antibacterial Effect of Honey on Bacterial Wound Pathogens, Journal of Applied Sciences Research, 6(11):1561-1565.
- [66] Ndife J., Abioye L. and Dandago M. (2014): Quality Assessment of Nigerian Honey Sourced from Different Floral Locations, Nigerian Food Journal, 32(2):48-55.
- [67] Nombre I., Schweitzer P., Boussim J., Millogo J. and Rasolodimby J. (2010): Impacts of Storage Conditions on Physico-Chemical Characteristics of Honey Samples from Burkina Faso. African Journal of Food Science 4(7):458-463.
- [68] Escuredo O., Seijo M. and Fernandez-Gonzalez M. (2011): Descriptive Analysis of Rubus Honey from the North-West of Spain. International Journal of Food Science and Technology 46:2329-2336.
- [69] Guo W., Liu Y., Zhu X. and Wang S., (2011): Temperature Dependent Dielectric Properties of Honey Associated with Dielectric Heating. Journal of Food Engineering, 102:209–216.
- [70] Baglio E. (2018): Chemistry and Technology of Honey Production. Molecular Science. Chem. Foods, 59:12-15.
- [71] Szilard B., Ioan T., Gyozo G., Ioana AV. and Florin D. (2020): Heavy Metal Content in Polyfloral Honey and Potential Health Risk. A Case Study of Copsa Mica, Romania. International Journal of Environmental Research and Public Health. 17(1507):1-12.
- [72] Blassa M., Candracci M., Accorsi A., Piacentini MP. and Albertini MC. (2006): Raw Mellefiori Honey is Packed Full of Antioxidants, Food Chemistry, 97:217-222.
- [73] Amabye TG. and Frehiwot M. (2016): Phytochemical and Biochemical Composition of Wild Honey a Case Study in Eastern Zone Areas in Tigray Ethiopia, Journal of Nutritional Health and *Food Engineering*, 4(4):487-492.



Dr. Yau Datti obtained his Ph.D in the year 2015 from Bayero University, Kano, Nigeria. He is currently working as a lecturer with Yusuf Maitama Sule University Kano (formerly Northwest University, Kano) Nigeria. He has published research papers in many reputed international journals. His main research work focuses on stress biology. His main areas of my research interest are organic synthesis, natural products and drug discovery. He has 12 years of teaching and research experience.



Ummi Umar Ahmad is currently working as a lecturer with School of Technology, Kano State Polytechnic, with many years of teaching experience. She has completed her Bachelor of Science in the field of Chemistry, while her Masters of Science is in the field of Analytical Chemistry. Currently, she is pursuing her PhD. In Analytical Chemistry in Bayero University, Kano, Nigeria. She has published research paper in many reputed journals.



Usman Lado Ali, born on 3rd March, 1982 at Getso Town, Gwarzo Local Government, Kano State, Nigeria. Graduated from Bayero University Kano, Nigeria and obtained B. Sc. Honor in Chemistry, 2006. Worked with KSSSSMB Kano, Science and TechnicalSchools Board, Kano and Currently graduate assistant at Northwest University Kano, as well as M.Sc. Chemistry Student at SRM University, Kattankulathur-603203, India.